organic papers

Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

Liang-Zhong Xu,* Kai Li, Chong-Yi Zhu, Guo-Dong Si and **Guan-Ping Yu**

Institute of Agricultural Chemicals, Qingdao University of Science and Technology, Qingdao 266042, People's Republic of China

Correspondence e-mail: gknhs@163169.net

Key indicators

Single-crystal X-ray study T = 295 K Mean σ (C–C) = 0.003 Å R factor = 0.034 wR factor = 0.086 Data-to-parameter ratio = 13.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(4-Fluorophenyl)-2-oxo-1-(1H-1,2,4triazol-1-yl)ethyl piperidine-1-carbodithiolate

The molecule of the title compound, $C_{16}H_{17}FN_4OS_2$, is built up around a chiral C atom, and exists in a propeller-like arrangement. The structure is stablized by van der Waals interactions.

Comment

As an important type of fungicides, triazole compounds are highly efficient and of low toxicity (Shi et al., 1995; Xu et al., 2002). Triazole nuclei appear frequently in the structures of various natural products and biologically active compounds, notably thiamine (vitamin B), penicillins, antibiotics such as micrococcin (James et al., 1966), and many metabolic products of fungi and primitive marine animals. Present studies of triazole derivatives concentrate mainly on compounds with triazole as the only active group, while reports of compounds containing both triazole and piperidine groups in a single molecule are scarce. We have therefore studied the title compound, (I), and present its structure here.



As shown in Fig. 1, the molecule of (I) is built up around the chiral atom C7, which is connected to a piperidine carbodithiolate, a triazole and a fluorophenyloxo group, in a propeller-like arrangement. The four atoms S1, S2, N1 and C1 (p1) of the carbodithiolate are planar, as are the triazole and fluorophenyl groups. The dihedral angles between p1 and the triazole and fluorophenyl groups are 66.69(8) and $81.89(7)^{\circ}$, respectively. The dihedral angle between the triazole and fluorophenyl groups is $67.88 (4)^{\circ}$.

The bond lengths and angles in the 1,2,4-triazole and phenyl rings are generally normal (Ji et al., 2002; Allen et al., 1987). The bond lengths and angles within the piperidine ring are also in good agreement with the earlier report by Yuan et al. (2004). The C-F bond length [1.357 (2) Å] is similar to that found by Lynch & McClenaghan (2004) [1.340 (3)-1.345 (3) Å].

There are some weak C-H···S intramolecular hydrogen bonds (Table 1) (Song et al., 2003). The packing is stabilized by weak van der Waals interactions.

Printed in Great Britain - all rights reserved 01306

Xu et al. • $C_{16}H_{17}FN_4OS_2$

© 2005 International Union of Crystallography

Received 14 March 2005 Accepted 5 April 2005 Online 9 April 2005

Experimental

The title compound was prepared by the reaction of 2-bromo-1-(4-fluorophenyl)-2-(1*H*-1,2,4-triazol-1-yl)ethanone (0.02 mol, 5.7 g), piperidine (0.02 mol, 1.7 g), CS₂ (0.02 mol, 1.5 g) and potassium hydroxide (0.02 mol, 1.1 g) in ethanol solution at room temperature. Single crystals of (I) suitable for X-ray measurements were obtained by recrystallization from chloroform at room temperature.

Crystal data

$C_{16}H_{17}FN_4OS_2$	Z = 2
$M_r = 364.46$	$D_x = 1.435 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 8.7112 (15) Å	Cell parameters from 1376
b = 9.8967 (17) Å	reflections
c = 11.1551 (19) Å	$\theta = 2.5 - 24.1^{\circ}$
$\alpha = 73.323 \ (2)^{\circ}$	$\mu = 0.34 \text{ mm}^{-1}$
$\beta = 88.603 \ (2)^{\circ}$	T = 295 (2) K
$\gamma = 66.981 \ (2)^{\circ}$	Block, colourless
V = 843.7 (3) Å ³	0.24 \times 0.18 \times 0.16 mm

Data collection

Bruker SMART CCD area-detector	2932 independent reflections
diffractometer	2231 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.015$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -9 \rightarrow 10$
$T_{\min} = 0.910, \ T_{\max} = 0.948$	$k = -11 \rightarrow 11$
4609 measured reflections	$l = -13 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0427P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.034$	+ 0.0282P]
$wR(F^2) = 0.086$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.10	$(\Delta/\sigma)_{\rm max} < 0.001$
2932 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$
217 parameters	$\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

Hydrogen-bond geometry (Å, °).

D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
0.97	2.53	3.052 (3)	114
0.97	2.39	2.939 (2)	115
0.98	2.45	3.108 (3)	124
0.93	2.81	3.735 (2)	172
	<i>D</i> -H 0.97 0.97 0.98 0.93	$\begin{array}{c ccc} D-H & H\cdots A \\ \hline 0.97 & 2.53 \\ 0.97 & 2.39 \\ 0.98 & 2.45 \\ 0.93 & 2.81 \end{array}$	$D-H$ $H \cdots A$ $D \cdots A$ 0.97 2.53 3.052 (3) 0.97 2.39 2.939 (2) 0.98 2.45 3.108 (3) 0.93 2.81 3.735 (2)

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C-H distances in the range 0.93–0.97 Å and with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$.



Figure 1

The structure of (I), showing 40% probability displacement ellipsoids and the atom-numbering scheme.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- James, M. N. G. & Watson, K. J. (1966). J. Chem. Soc. pp. 1361-1367.
- Ji, B. M., Du, C. X., Zhu, Y. & Wang, Y. (2002). Chin. J. Struct. Chem. 21, 252– 255.
- Lynch, D. E. & McClenaghan, L. (2004). Acta Cryst. C60, 01-05.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Shi, Y. N., Lu, Y. C. & Fang, J. X. (1995). Chem. J. Chin. Univ. 16, 1710–1713.
- Song, Q.-B., Li, Y.-Z., Liang, Y.-M. & Ma, Y.-X. (2003). Acta Cryst. E59, m861– m863.
- Xu, L. Z., Zhang, S. S., Li, H. J., Jiao, K. (2002). Chem. Res. Chin. Univ. 18, 284–286.
- Yuan, D., Zhang, M., Pan, Z. & Ma, P. (2004). Acta Cryst. E60, o1321-o1322.